

## Paleotemperatures

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### Definition

A paleotemperature is the temperature of a location, either on land or in the ocean, at a specific time in the geologic past. Scientists use paleothermometers to reconstruct past temperature records from archives – natural features of the Earth that preserve clues about past climate and environmental change. Examples of archives include marine and lacustrine sediment, glacial ice, and corals. Paleothermometers are often referred to as temperature proxies, which are measurable physical, geological, geochemical, or biological characteristics that get preserved in archives, standing in for instrumental measurements. Each proxy is associated with a calibration that transforms the proxy measurements into the parameter of interest, which in this case is temperature.

Scientists rely on proxy measurements from archives because direct observations of temperature extend back only as far as the middle of the nineteenth century (Smith et al. 2008). Direct temperature measurements of the ocean extend back even less. These limits to the length of instrumental records make it difficult to constrain the range of natural climate variability, the degree of twentieth century warming, and climate sensitivity to greenhouse gas forcing. Accurate paleotemperature reconstructions are the only way to extend the short observational period farther into the past.

This entry describes paleotemperature proxies from marine and terrestrial archives. The paleotemperature proxies presented here are by no means an exhaustive list, but instead

highlight the geochemical proxies that are used most frequently for paleotemperature reconstructions.

### Assemblages of Planktonic Foraminifera

Foraminifera are marine protozoa (unicellular zooplankton) that can either be found free-floating in the open, upper layers of the ocean (planktonic), or living on or in seafloor sediment (benthic). They secrete characteristic tests, or shells, often composed of calcium carbonate (calcite,  $\text{CaCO}_3$ ), that accumulate on the seafloor after death. There are ~50 living species of planktonic foraminifera in the modern oceans, and they are found in diverse oceanic regions from the warm tropics to cold polar waters (Kucera 2007). The greatest planktonic species diversity can be found in the tropics, with diversity generally decreasing with increasing latitude. Differences in the species composition, or assemblage, of planktonic foraminifera contain information about the temperature of the waters they inhabit (Murray 1897). Statistical techniques can be used to determine sea surface temperature (SST) from assemblages, including transfer functions (Imbrie and Kipp 1971) and the modern analog technique (Hutson 1980; Howard and Prell 1992). Studies comparing both methods find little difference between them (Ortiz and Mix 1997), but there are limitations to the overall approach. Both methods are based on the assumption that temperature is the main control on the assemblages; other factors, such as food availability, light requirements, and ocean circulation changes can also alter species distributions (Wefer et al. 1999). Finally, there is uncertainty in relating habitat (depth and seasonal) to SST at a certain time of the year.

### Mg/Ca Ratios

Mg/Ca ratios in benthic and planktonic foraminifera have become widely used proxies for reconstructing bottom and

upper water column paleotemperatures, respectively. The underlying basis for Mg/Ca paleothermometry is that the substitution of  $\text{Mg}^{2+}$  for  $\text{Ca}^{2+}$  in calcite is endothermic and therefore favored at higher temperatures. As a result, the Mg/Ca ratio of foraminiferal calcite increases with increasing temperature.

The temperature dependence on Mg uptake into planktonic foraminiferal tests has been determined using culture-based, sediment trap, and core-top calibrations. Culture-based calibrations rely on growing planktonic foraminifera under controlled laboratory conditions in which the temperature is fixed independently of other environmental parameters (e.g., Nürnberg et al. 1996; Lea et al. 1999; Russell et al. 2004; Allen et al. 2016). Sediment trap calibrations measure planktonic foraminifera collected in sediment trap time series from regions with high seasonal temperature variability (e.g., Anand et al. 2003). Core-top calibrations measure Mg/Ca ratios of modern or recently deposited planktonic foraminifera from the seafloor and then calibrate the ratios against the water column temperature overlying the sediment core site based on the foraminifera's preferred depth habitat (e.g., Lea et al. 2000; Dekens et al. 2002). Core-top calibrations are used most often to determine the relationship between Mg/Ca ratios in benthic foraminifera and bottom water temperature (Rosenthal et al. 1997; Lear et al. 2002; Bryan and Marchitto 2008).

Despite differences between calibration methods, studies show that most species of planktonic foraminifera have a temperature sensitivity of a ~9–10% exponential increase in Mg/Ca per degree Celsius (Lea et al. 1999; Elderfield and Ganssen 2000; Dekens et al. 2002; Anand et al. 2003). Culturing studies find that salinity also influences the uptake of Mg into planktonic foraminiferal tests, with Mg/Ca ratios increasing by 3–8% per salinity unit increase (Nürnberg et al. 1996; Lea et al. 1999; Hertzberg and Schmidt 2013; Hönisch et al. 2013; Allen et al. 2016). In contrast, the sensitivity of Mg/Ca to temperature in benthic foraminifera is more variable between species and may be influenced by other factors, including bottom water carbonate ion concentration (Lear et al. 2002; Elderfield et al. 2006; Bryan and Marchitto 2008). As benthic foraminifera dwell on the bottom of the ocean, where temperatures are lower than at the sea surface, their Mg/Ca ratios tend to fall on the lower end of the exponential relationship with temperature. Thus, the Mg/Ca paleothermometer for benthic foraminifera is less sensitive than for planktonic foraminifera, often making bottom water Mg/Ca-temperature estimates more difficult.

Postdepositional processes can alter Mg/Ca ratios in planktonic foraminifera because saturation with respect to calcite decreases in the ocean with increasing depth. Depth transect studies show that, in general, Mg/Ca ratios decrease with increasing water depth, especially below the regional lysocline (Brown and Elderfield 1996; Lea et al. 2000;

Dekens et al. 2002; Regenberg et al. 2014). Partial dissolution of foraminifera tests can bias Mg/Ca ratios due to the heterogeneous distribution of Mg within the test and the higher solubility of Mg-rich calcite (Brown and Elderfield 1996; Dekens et al. 2002). However, methods have been developed to account for this effect (Rosenthal and Lohmann 2002; Dekens et al. 2002; Regenberg et al. 2014).

Recent advances in Mg/Ca paleothermometry include the analysis of multiple planktonic species that live at a range of depths and the analysis of individual foraminiferal tests from the same stratigraphic interval. By analyzing species with a range of depth habitats, scientists can reconstruct past changes in the vertical temperature profile of the ocean (Schmidt et al. 2012; Parker et al. 2015). To gain insight into short-term variability at times in the past, scientists are now using individual planktonic foraminifera Mg/Ca analyses from a single depth horizon in a sediment core (Ford et al. 2015). This approach allows scientist to reconstruct past changes in overall temperature variability and has provided insight into past changes in seasonality and El Niño Southern Oscillation variability.

Mg/Ca ratios in ostracods can be also used to reconstruct bottom water temperatures (e.g., Cronin et al. 2000). Ostracods are benthic bivalved Crustacea that secrete calcite shells during molting. Similar to foraminifera, the amount of magnesium in the calcite shells is largely controlled by the water temperature in which the organism secreted the shell (Dwyer et al. 2002). A positive correlation between water temperature and Mg/Ca in ostracods has been observed, although post-depositional processes such as dissolution also impact Mg/Ca ratios (Cronin et al. 2000). The main difference between foraminifera and ostracod Mg/Ca paleothermometry is that the calibration between temperature and Mg/Ca in ostracods is linear, as opposed to exponential for foraminifera.

## Oxygen Isotopic Composition of Foraminifera

The oxygen isotopic composition of both planktonic and benthic foraminiferal calcite has been used for decades as a proxy for reconstructing past SSTs (Emiliani 1955). It is expressed as  $\delta^{18}\text{O}_{\text{Calcite}}$  ( $\delta^{18}\text{O}_{\text{C}}$ ), the deviation of the  $^{18}\text{O}/^{16}\text{O}$  ratio relative to the VPDB standard in per mil (‰). Based on laboratory and natural experiments, a number of general and species-specific foraminiferal relationships between SST and  $\delta^{18}\text{O}_{\text{C}}$  have been published (Epstein et al. 1953; Shackleton 1974; Erez and Luz 1983; Kim and O'Neil 1997; Bemis et al. 1998; Marchitto et al. 2014). In general,  $\delta^{18}\text{O}_{\text{C}}$  decreases by ~0.21–0.23 ‰ for a 1 °C increase in temperature (Ravelo and Hillaire-Marcel 2007). However, foraminiferal  $\delta^{18}\text{O}_{\text{C}}$  values are a function of both temperature and the oxygen isotopic composition of the seawater ( $\delta^{18}\text{O}_{\text{SW}}$ ) in which they grow. Furthermore,  $\delta^{18}\text{O}_{\text{SW}}$  varies

with changes in salinity, as evaporation and precipitation cause local  $\delta^{18}\text{O}_{\text{SW}}$  to increase and decrease, respectively. On longer timescales, changes in continental ice volume cause  $\delta^{18}\text{O}_{\text{SW}}$  to decrease globally with melting of ice and increase globally with ice sheet growth. Mixing of water masses with different  $\delta^{18}\text{O}_{\text{SW}}$  values can also influence local  $\delta^{18}\text{O}_{\text{SW}}$ . Thus, only if  $\delta^{18}\text{O}_{\text{SW}}$  is known (or at least constrained) can  $\delta^{18}\text{O}_{\text{C}}$  provide unique paleotemperatures. Other limitations to the  $\delta^{18}\text{O}_{\text{C}}$  temperature proxy are vital effects such as photosynthetic activity of algal symbionts (Duplessy et al. 1970; Bemis et al. 1998), changing depth habitats of planktonic foraminifera, and the carbonate ion concentration of seawater (Spero et al. 1997).

Because  $\delta^{18}\text{O}_{\text{C}}$  is a function of both temperature and  $\delta^{18}\text{O}_{\text{SW}}$ , researchers developed a method using paired Mg/Ca temperatures and  $\delta^{18}\text{O}_{\text{C}}$  measurements on foraminifera to reconstruct past  $\delta^{18}\text{O}_{\text{SW}}$  variability. When applied to surface-dwelling planktonic foraminifera, this method has been shown to be a robust proxy for reconstructing sea surface salinity variability once changes in continental ice volume are taken into account (Schmidt et al. 2006; Schmidt and Lynch-Stieglitz 2011). Researchers have also used this method with benthic foraminifera to reconstruct bottom water  $\delta^{18}\text{O}_{\text{SW}}$  change as a proxy for continental ice volume variability over the past 1.5 Ma (Elderfield et al. 2012). The most significant advantage of this approach to calculate  $\delta^{18}\text{O}_{\text{SW}}$  variability is that Mg/Ca can be measured on the same shells that carry  $\delta^{18}\text{O}_{\text{C}}$  information, so there is no ambiguity in timing between the signals.

### Clumped Isotopes in Carbonates

Recently, a considerable amount of interest has been invested in the development of a paleothermometer based on clumped stable isotope analyses of  $\text{CaCO}_3$ . Although the multiply substituted isotopologue ( $^{13}\text{C}^{16}\text{O}^{18}\text{O}$ ) is rare and difficult to measure, the approach is based on the thermodynamic prediction that the abundance of multiple isotope substitutions in  $\text{CaCO}_3$  increases as temperature decreases (Schauble et al. 2006; Passey et al. 2010). Therefore, unlike for temperature reconstructions based on  $\delta^{18}\text{O}$  values, temperature reconstructions based on this method are not influenced by changes in the  $^{18}\text{O}/^{16}\text{O}$  ratio of the water from which the mineral was precipitated. Initially, synthetically grown carbonates were used to identify the relationship between temperature and  $\Delta_{47}$  values (Ghosh et al. 2006). Since then, calibration efforts have extended to shallow and deepwater corals, mollusks, and fish otoliths (Ghosh et al. 2006, 2007; Came et al. 2007; Thiagarajan et al. 2011). In addition, recent work has included the initial development of clumped isotopes in foraminifera and coccolithophores (Tripathi et al. 2010, 2014). The resulting

relationship for both inorganic and organic calcite shows a sensitivity of  $\sim 0.004\text{--}0.005\text{‰ per }^\circ\text{C}$ .

Although clumped isotopes are becoming a more widely applied tool for geosciences, its application to the field of paleoceanography has been somewhat limited because of several factors. First, large sample sizes are required for a single analysis (in the range of 3–8 mg  $\text{CaCO}_3$ ). Second, because the measurement of  $\Delta_{47}$  is analytically challenging, it is necessary to replicate each measurement three to five times in order to reduce the standard error to less than  $\pm 2^\circ\text{C}$ . Third, only limited calibration data is available, especially for temperatures below  $15^\circ\text{C}$ .

The abundance of  $^{13}\text{C}\text{--}^{18}\text{O}$  bonds in a  $\text{CaCO}_3$  sample is measured on the  $\text{CO}_2$  liberated from the sample in phosphoric acid using gas source mass spectrometry (Ghosh et al. 2006; Guo et al. 2009). This allows for both  $\Delta_{47}$  and  $\delta^{18}\text{O}$  analysis on the same sample, resulting in the determination of both calcification temperature and the  $\delta^{18}\text{O}$  value of the water from which the mineral precipitated. Therefore, as analytical methods continue to improve, this technique has considerable potential for becoming an important paleothermometer for future paleoceanographic studies.

### Biomarker Proxies: Alkenones and $\text{TEX}_{86}$

Although the Mg/Ca paleothermometer has advantages compared with the  $\delta^{18}\text{O}_{\text{C}}$  proxy, it also has limitations including the influences of dissolution, calibrations based only on modern species, geographic limitations to regions inhabited by foraminifera, and temporal variability in seawater Mg/Ca ratios over millions of years. Biomarker-based SST proxies, such as the alkenone  $\text{U}^k_{37}$  (Brassell et al. 1986; Prah and Wakeham 1987) and  $\text{TEX}_{86}$  (Schouten et al. 2002) indices, can overcome some of these limitations due to the ubiquitous appearance of the associated biomarkers in marine sediments temporally and spatially, especially during times and in regions where foraminifera do not exist. Similar to foraminifera-based proxies, the carriers of biomarkers accumulate and get preserved on the seafloor and can be sampled and extracted from sediment cores.

Alkenones are long-chain organic compounds produced by marine haptophyte algae (phytoplankton) known as coccolithophores. Because algae require sunlight to perform photosynthesis, they reside at or near the sea surface, making them a useful tool for reconstructing SST. The  $\text{U}^k_{37}$  index is based on the proportions of di- and tri-unsaturated ketones within the alkenones and varies positively with growth temperature. A decrease in temperature leads to an increase in the degree of unsaturation. Prah and Wakeham (1987) and Prah et al. (1988) developed the first quantitative calibration of alkenone unsaturation to growth temperature based on coccolithophores grown in the laboratory at known

temperatures, and it appeared to reliably estimate SST based on samples collected from the ocean. Since then, large datasets of core-top  $U^{k'}_{37}$  SST calibrations have been developed and show strong convergence with the original laboratory based calibration (Conte et al. 2006). Unfortunately, one of the main drawbacks to the  $U^{k'}_{37}$  paleotemperature proxy is that it has an upper temperature limit of  $\sim 28$  °C and a lower limit of  $\sim 1$  °C (Herbert 2014). This precludes SST reconstructions based on the  $U^{k'}_{37}$  index from the coldest and warmest regions of the ocean and from exceptionally warm periods in Earth's history.

Other ambiguities related to the alkenone paleotemperature proxy are depth and seasonal variations of alkenone-producing species in the ocean (Herbert 2014) and the potential for postdepositional transport of fine-grained alkenone carriers (Ohkouchi et al. 2002). Sunlight can penetrate to depths below the sea surface; thus, alkenones can be produced at a range of depths as long as light is sufficient for photosynthesis. Another drawback to this proxy is that phytoplankton production is seasonal in most ocean locations, so the  $U^{k'}_{37}$  temperatures can be biased to the season of maximum production.

The  $TEX_{86}$  paleothermometer is another biomarker-based proxy developed more recently and holds promise for overcoming the temperature limitations of the alkenone temperature proxy. The  $TEX_{86}$  (TetraEther indeX of 86 carbons) index is based on the number of cyclopentane rings in membrane lipids, known as glycerol dialkyl glycerol tetraethers (GDGTs), which are produced by marine Thaumarchaeota (Schouten et al. 2002). The  $TEX_{86}$  paleothermometer is based on observations from culturing and mesocosm experiments that the number of cyclopentane rings in GDGTs increases with increasing growth temperature (Wuchter et al. 2004). Schouten et al. (2002) noted a relationship between the number of rings in the GDGTs in marine surface samples and overlying SST and developed the first  $TEX_{86}$  temperature calibration based on 40-sediment core-top samples from 15 locations. Subsequent calibrations involving the addition of hundreds of core-top samples all note the highest correlation between the  $TEX_{86}$  index and SST (Kim et al. 2010). However, other studies have noted a bias in  $TEX_{86}$  temperatures where they are representative of subsurface rather than sea surface temperatures (Hertzberg et al. 2016). Unlike coccolithophorids that live near the sea surface, archaea live throughout the water column, which may help explain the subsurface signal (Schouten et al. 2002). While the exact cause of the subsurface bias is still unknown, a new calibration to subsurface temperature has been developed (Tierney and Tingley 2015). Despite this uncertainty, the  $TEX_{86}$  temperature proxy, which can reconstruct warmer temperatures than the alkenone SST proxy, has been used to reconstruct SSTs from the early Eocene when atmospheric  $pCO_2$  levels may have reached  $\sim 2000$  ppmv (Bijl et al. 2009).

The  $TEX_{86}$  paleothermometer can also be applied to lacustrine sediments for reconstructions of terrestrial temperatures, as GDGT-producing Thaumarchaeota also exist in lakes (Powers et al. 2004). The mechanistic basis of the proxy is the same in these systems, although separate calibrations are necessary (Tierney et al. 2010). In addition, soil GDGTs and methanogenic archaea can potentially complicate  $TEX_{86}$  in lakes, but these effects do not appear to be systematic (Tierney 2014). Nevertheless, the lacustrine  $TEX_{86}$  paleothermometer has been applied to a number of lakes, in, for example, Africa (Tierney et al. 2008).

### Sr/Ca Ratios in Corals

Corals associated with reefs (as opposed to deep-sea corals) grow in regions of warm SST making them a useful archive for tropical SST reconstructions. These corals secrete aragonite in annual bands, so high-resolution paleotemperature reconstructions are possible.  $Sr^{2+}$  substitution in corals has inverse temperature dependence due to the exothermic nature of the reaction in which  $Sr^{2+}$  substitutes for  $Ca^{2+}$  in aragonite. Typically, Sr/Ca ratios are first calibrated with modern instrumental SST records to establish a coral-specific Sr/Ca-temperature relationship, which is then applied to the older skeleton of the same coral, or to fossil corals for reconstructions past SST (e.g., Felis et al. 2009; Hereid et al. 2013).

Coral Sr/Ca ratios appear to also be strongly influenced by vital effects (Alpert et al. 2016). Growth rate and symbiotic activity have a marked influence on coral Sr/Ca, and culture studies have shown that coral Sr/Ca responds to changes in seawater pH and light. Researchers have also noted offsets in temperature sensitivity and mean Sr/Ca ratios in neighboring colonies collected from the same reef (Goodkin et al. 2005; Alpert et al. 2016). A potential explanation for these vital effects is that corals accrete their skeleton within an isolated calcifying fluid that is continuously depleted in Sr/Ca as aragonite precipitation proceeds (Cohen et al. 2006). Changes in the carbonate ion concentration of the calcifying fluid can drive variations in the amount of aragonite precipitation and cause variations in the magnitude of calcifying fluid Sr/Ca depletion. DeCarlo et al. (2016) developed a method to account for these vital effects by combining Sr/Ca with U/Ca ratios, which decrease as carbonate ion concentrations increase. This new method, termed coral Sr-U thermometry, has the potential to offer significantly improved reliability for reconstructing past ocean temperatures from corals.

### Terrestrial Temperature Proxies

Terrestrial temperature proxies provide information on Earth's surface air temperature. Geochemical proxies such

as oxygen ( $\delta^{18}\text{O}$ ) and deuterium ( $\delta\text{D}$ ) isotopic values of glacial ice, and nitrogen and argon isotopic ratios of air bubbles trapped in ice, can be used to estimate past atmospheric temperatures.

Ice cores are a valuable paleoclimate archive because of the large number of variables that can be derived from both the ice and the air that is trapped within the ice as it forms. In addition to temperature, ice cores are used for reconstructions of atmospheric  $\text{CO}_2$  and  $\text{CH}_4$  concentrations, dust, and concentrations of salts, to name a few. Ice cores can be drilled from the polar regions of Greenland and Antarctica and from high-altitude tropical and subtropical locations. Oxygen and deuterium isotopes measured in ice can be used as a temperature proxy because isotopic fractions of the heavier  $^{18}\text{O}$  and  $^2\text{H}$  in snowfall are temperature dependent, and there is a strong spatial correlation between mean annual temperature and the mean isotopic fraction of  $^{18}\text{O}$  or  $^2\text{H}$  in precipitation (Dansgaard et al. 1993; Grootes et al. 1993). Evaporation fractionation, which depends on surface conditions (e.g., SST, wind speed, relative humidity), controls the isotopic composition of atmospheric water vapor. Fractionation occurs as an air mass moves poleward, cooling and condensing, which leads to fractionation as the air mass progressively loses heavy isotopes from the water vapor. As a result of this process, known as Rayleigh distillation, the predominant control on the polar isotopic composition of water vapor is the temperature difference between the evaporative source and the final region of deposition (Masson-Delmotte, et al. 2006). Thus, past changes in evaporation conditions or atmospheric transport can induce biases in temperature reconstructions based on  $\delta^{18}\text{O}$  or  $\delta\text{D}$ . A parameter known as deuterium excess ( $d = \delta\text{D} - 8(\delta^{18}\text{O})$ ) takes advantage of the combined measurements of  $\delta^{18}\text{O}$  and  $\delta\text{D}$  to correct past temperature reconstructions from changes in evaporation conditions (Cuffey and Vimeux 2001). Prominent ice core temperature records from Greenland include GRIP (Greenland Ice-core Project Members 1993; Dansgaard et al. 1993), GISP2 (Grootes et al. 1993), and NGRIP (North Greenland Ice Core Project members 2004) and from Antarctica include Vostok (Petit et al. 1999), EPICA Dome C (EPICA community members 2004), and WAIS Divide (WAIS Divide Project Members 2013). Note that this is not a complete list of all available ice core records.

Measurements of the  $\delta^{15}\text{N}$  and  $\delta^{40}\text{Ar}$  of air bubbles trapped in ice cores can be used for temperature reconstructions during periods of abrupt temperature change. When rapid temperature change occurs, a transient temperature gradient is established in the firn (the region of partially compacted snow) until a stable depth temperature gradient in the upper part of the ice is restored. Under a vertical temperature gradient, the gases in the firn undergo a thermal fractionation, with the heaviest species being concentrated at the coldest end (Masson-Delmotte et al. 2006). Analyses of

$\delta^{15}\text{N}$  and  $\delta^{40}\text{Ar}$  in ice core air bubbles, which are constant in the atmosphere over long timescales, can be combined with firn modeling to provide estimates of past abrupt temperature changes (Severinghaus et al. 1998). This method has proven especially useful for reconstructing temperature changes from Greenland ice cores across abrupt climate change intervals such as Dansgaard-Oeschger events of the last glacial period and the millennial-scale events of the last deglaciation (Severinghaus and Brook 1999; Landais et al. 2004).

## Conclusions

Each paleotemperature proxy has a unique set of advantages and disadvantages. Depending on the location and time period of interest, scientists will use the proxies that best suit their needs. Often, reconstructing temperature using two or more proxies can help to make climate records more robust. If the paleotemperatures measured using multiple proxies disagree, it may indicate an environmental control on the proxy unrelated to temperature. Finally, robust and reliable age models are key to any paleotemperature reconstruction. Temperature proxies also continue to undergo refinement as new discoveries are made and better technologies and methods are made available for measurements. The geochemical paleotemperature proxies discussed in this entry are by no means an exhaustive list, and the reader is encouraged to consult the references for more details.

## Cross-References

- ▶ [Carbonate Sediments](#)
- ▶ [Oxygen Isotopes](#)
- ▶ [Paleoclimatology](#)
- ▶ [Paleoenvironments](#)
- ▶ [Paleoproductivity](#)
- ▶ [Stable Isotope Geochemistry](#)

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